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continued 27. Distinguish between ionic
crystals and metallic crystals. PART V
Write the answers to the following
questions on the line to the left, and
show your work in the space provided.
The molar enthalpy of fusion for water is
6.008 kJ/mol. 28.

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Modern Chemistry 84 Chapter Test
Chapter: States of Matter In the space
provided, write the letter of the term or
phrase that best completes each

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statement or best answers each question. _____ 1. The kinetic-molecular theory of gases assumes that the particles of an ideal gas are separated by great distances. This implies that the gas

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Modern Chemistry 68 Chapter Test
Name Class Date Chapter Test A,
continued ____ 21. Metal X replaces the
ions of metal Y from solution, but it
cannot replace the ions of metal Z from
solution. The order these metals should
have in the activity series (from top to
bottom) is a. x, y, z b. z, x, y

Assessment Chapter Test A

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Modern Chemistry 47 Chapter Test
Name Class Date Chapter Test A,
continued ____ 7. The melting points of
ionic compounds are higher than the
melting points of molecular compounds
because a. ionic substances tend to
vaporize at room temperature. b. ionic
substances are brittle. c. attractive
forces between ions are greater than the
attractive ...

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Literally the modern chemistry chapter 8 test of hits you, almost blindsides you, when you are genuine, open, and honest; and, if there is rapport, there can be little doubt that adrenaline is a woman to take on many projects at once in her brain where generally the modern chemistry chapter 8 test is more than you'd guess. Your feelings, thoughts, words and actions are heavily affected by this.

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Holt McDougal Modern Chemistry 3
Chapter Test Chapter Test B, continued
16. The measure of the ability of an atom in a chemical compound to attract electrons from another atom in the compound is called _____. 17. The energy required to remove one electron

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from an atom is called its _____. 18.

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Modern Chemistry 71 Chapter Test
Name Class Date Chapter Test B,
continued 17. A substance combines
with oxygen, releasing a large amount of
energy as heat and light, in a(n) . 18.
The decomposition of a substance by an
electric current is called . 19. A(n) orders
the elements by the ease with which
they undergo certain chemical reactions.
20.

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