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Name Class Date Chapter Test B,  
continued 31. Calculate the energy  
released by freezing 13.8 g of a liquid.  
Its molar mass is 82.9 g/mol, and its  
molar enthalpy of fusion is 4.60 kJ/mol.  
PART VI Define the term equilibrium  
vapor pressure. Use it to explain the  
following two phenomena.

### **Assessment Chapter Test A**

Modern Chemistry 33 Chapter Test  
Name Class Date Chapter Test B,

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continued 15. The energy state of an atom is called its ground state. 16. The number of waves that pass a point in one second is called. 17. When an electron drops from a higher-energy state to a lower-energy state, a(n) spectrum is produced. 18.

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This scientist proposed to explanation of the photoelectric effect is electromagnetic radiation in whole numbers of photons and in order for an electron to be ejected ...

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chapter of this Holt McDougal Modern Chemistry Companion Course helps students learn the essential lessons associated with how electrons are arranged in atoms.

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Modern Chemistry 71 Chapter Test  
Name Class Date Chapter Test B,  
continued 17. A substance combines  
with oxygen, releasing a large amount of  
energy as heat and light, in a(n) . 18.  
The decomposition of a substance by an  
electric current is called . 19. A(n) orders  
the elements by the ease with which  
they undergo certain chemical reactions.  
20.

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Holt McDougal Modern Chemistry 1 Chapter Test Assessment Chapter Test B Chapter: The Periodic Law PART I In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question. \_\_\_\_\_ 1. In his periodic table, Mendeleev did not list all of the elements in order

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## **New Page 1 [srvhs.org]**

Modern Chemistry 111 Chapter Test  
Name Class Date Chapter Test B,  
continued 18. Dissolution processes with  
negative enthalpies of solution are  
processes. 19. is the solution process  
with water as the solvent. 20. Liquid  
solutes and solvents that are not soluble  
in each other are. 21. A solute molecule  
that is surrounded by solvent molecules  
is ...

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Modern Chemistry Chapter 4 Test.  
Section 1-3. STUDY. PLAY. Orbital. a  
3-dimensional region around the nucleus  
that indicates the probable location of  
an electron. ... Modern Chemistry  
Chapter 4. 26 terms. Modern Chemistry  
Chapter 4. 70 terms. Holt Modern  
Chemistry: Chapter 4 Test. 45 terms.  
Modern Chemistry Chapter 4. Features.

## **Assessment Chapter Test A - Ed W.**

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## **Clark High School**

Modern Chemistry 47 Chapter Test

Name \_\_\_\_\_ Class \_\_\_\_\_ Date \_\_\_\_\_ Chapter Test A,  
continued \_\_\_\_\_ 7. The melting points of ionic compounds are higher than the melting points of molecular compounds because a. ionic substances tend to vaporize at room temperature. b. ionic substances are brittle. c. attractive forces between ions are greater than the attractive ...

## **Modern Chemistry Chapter 4 Test Answer Key**

CHAPTER 4: ARRANGEMENT OF ELECTRONS IN ATOMS The following pages contain the bulk (but not all) of the information for the chapter 4 test. Focus on this content, but make sure to review class notes, activities, handouts, questions, etc. If you study this document and NOTHING else, you should at least be able to PASS the test.

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Modern Chemistry Chapter 4. STUDY. Flashcards. Learn. Write. Spell. Test. PLAY. Match. Gravity. Created by. dvann123. Arrangements of Electrons in Atoms. Terms in this set (45) electromagnetic radiation. A form of energy that exhibits wavelike behavior as it travels through space ( $3.00 \times 10^8$  m/s) ... most have 4 lobes; and their destiny runs ...

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Modern Chemistry 27 Chapter Test

Name \_\_\_\_\_ Class \_\_\_\_\_ Date \_\_\_\_\_ Chapter Test A,

continued \_\_\_\_\_ 4. The electron

configuration below violates a. the Pauli

exclusion principle. b. the Aufbau

principle. c. Hund's rule. d. Both (a) and

(c) \_\_\_\_\_ 5. A photon is emitted from a

gaseous atom when an electron moves

to its ground state from a(n) a. inner

shell. b ...