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Chapter 11 Modern Atomic Theory

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Assessment Chapter Test B

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Modern Chemistry • CHAPTER 11 HOMEWORK 11-2 (pp. 335–339) VOCABULARY Write true or false.

1. The standard molar volume of a gas is the volume occupied by one liter of gas at standard temperature and pressure. ____ 2. The standard molar volume of a gas is 22.4 L. ____ 3. In addition to having a standard molar volume, all gases have a ...

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CHAPTER 11 HOMEWORK 11-1

Force Pressure 1 atm of pressure is the force of 10.1 N on 1 cm² Oxygen molecule, O₂ Nitrogen molecule, N₂ Pressure One way to think of force is to consider it the result of a mass times an acceleration. A newton is the force that will increase the speed of a one-kilogram mass by one meter per

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11. An atom is said to be in an excited state when it possesses more than its minimum energy (its ground state). An atom is promoted from its ground state to an excited state by the absorption of energy, and returns to its ground state by the emission of the excess energy. 12.

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the rates of effusion of gases at the same temperature and pressure are inversely proportional to the square roots of their molar masses Rate of effusion of A / B = $\sqrt{M_b} / \sqrt{M_a}$

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Holt McDougal Modern Chemistry 1 Chapter Test Assessment Chapter Test B Chapter: The Periodic Law PART I In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question. ____ 1. In his periodic table, Mendeleev did not list all of the elements in order

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