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SCIENTIFIC NOTATION Name Scientists very often deal with very small and very large numbers, which can lead to a '\*' confusion when counting zeros! We have learned to express these numbers as powers  $\times 10^n$ .

## **SIGNIFICANT FIGURES Name**

When numbers in scientific notation are divided, only the number is divided. The exponents are subtracted.!

!  $10^7 \times 4$  # 6.00  
!  $10^3 \times 9.60$  !  $10^7 \times 1.60$  !  $10^4 \times 9.60 \times 1.60$  # Name Date Class

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96 Chemistry: Matter and Change Math Handbook Transparency Worksheets 1. For each of the measurements in the table below, determine if the underlined number is significant or not significant. Place a check mark in the appropriate box and in the box under the rule that you used to make your determination. 2.

## **Scientific Notation Use with Appendix B, Scientific**

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## Notation

Math Handbook Transparency Worksheets Name. 55. Scientists very often deal with very small and very large numbers, which can lead to a Pf confusion when counting zeros! We have learned to express these numbers as powers 10. 0 and 'n" represents the number Scientific notation takes the form of  $M \times 10^n$  where  $1 \leq M < 10$  and  $n$  is the number of decimal places to be moved.

## Math Handbook Transparency Worksheet

Math Handbook Transparency Masters Chemistry: Matter and Change 91 Scientists need to express small measurements, such as the mass of the proton at the center of a hydrogen atom

## Name Date Class 53 MATH HANDBOOK TRANSPARENCY MASTER 1 ...

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Companies, Inc. 92 Chemistry: Matter and Change Math Handbook Transparency Worksheets 1. Express each of the following numbers in scientific notation. ... Perform the following operations and express the answers in scientific notation. a ...

### **Pictures Math Handbook Transparency Worksheet, - Easy**

...

Math Handbook Transparency Masters Chemistry: Matter and Change 93 Addition and Subtraction Before numbers in scientific notation can be added or subtracted, the exponents must be equal. (3.4 10<sup>2</sup>) (4.57 10<sup>3</sup>) (0.34 10<sup>3</sup>) (4.57 10<sup>3</sup>) Not equal Equal The decimal is moved to the left to increase the exponent. (0.34 4.57) 10<sup>3</sup> 4.91 10<sup>3</sup>

### **snow.dearbornschools.org**

The answer is expressed as 190,000 cm<sup>3</sup> since 19 cm has only significant figures. When adding and subtracting, limit and round



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your answer to the least number of decimal places in any of the numbers that make up your answer. ... Math Handbook Transparency Worksheets . Author:

### **Math Skills Transparency Worksheet Answers Chapter 3**

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Math Handbook Transparency Worksheets Name Date Class Use with Appendix B, Scientific Notation MATH HANDBOOK TRANSPARENCY MASTER Scientific Notation Scientists need to express small measurements, such as the mass of the proton at the center of a hydrogen atom (0.000 000 000 000 000 000 000 000 001 673 kg), and large

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MATH SKILLS TRANSPARENCY WORKSHEET Determining  
Numbers of Ions Use with Chapter 7, Section 7.3

1. Write a simple word equation that illustrates what must be true of total positive charge and total negative charge in an ionic compound.
2. How many potassium ions (group 1) would be related to balance the charge of each of