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Chapter 9: Standard Review

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SECTION 3 PROBLEMS Write the answer
on the line to the left. Show all your work
in the space provided. 1. 88% The actual
yield of a reaction is 22 g and the
theoretical yield is 25 g. Calculate the
percentage yield. 2. 6.0 mol of N_2 are
mixed with 12.0 mol of H

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REVIEW. New York,, Jugoslovenski
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Modern Sector," LOCAL GOVERNMENT
THROUGHOUT THE WORLD, The Hague,
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-- (This is an answer to World History:
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1. Answers will vary. An example is included below: $2\text{H}_2\text{O}_2(\text{aq}) \rightarrow 2\text{H}_2\text{O}(\text{l}) + \text{O}_2(\text{g})$ This describes the decomposition reaction of hydrogen peroxide.

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stoichiometry answers**

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SHORT ANSWER Answer the provided a
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6 Chemical Bonding

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CHAPTER 6 REVIEW Chemical Bonding

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SECTION 4 SHORT ANSWER Answer the following questions in the space provided. 1. b In metals, the valence electrons are considered to be (a) attached to particular positive ions. (c) immobile. (b) shared by all surrounding atoms. (d) involved in covalent bonds.

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