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5 The Periodic Law

Modern Chemistry Chapter 7 Section 3 Flashcards | Quizlet. Modern Chemistry Chapter 7 Section 3. STUDY. PLAY. Formula Masses. any molecule, formula unit, or ion is the sum of the average atomic masses of all atoms represented in its formula. a compound's molar mass is numerically ... Modern Chemistry Chapter 7 Section 1 Review. 4 terms.

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(A) they did not like Marie Antoinette (B) they wanted to help the poor (C) the other two estates had privileges which were denied them (D) they supported the American Revolution

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CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds SECTION 2 SHORT ANSWER Answer the following questions in the space provided. 1. Assign the oxidation number to the specified element in each of the following examples: 4 a. S in H 2SO 3 6 b. S in MgSO 4 2 c. S in K 2S 1 d. Cu in Cu 2S 6 e. Cr in Na 2CrO 4 5 f. N in HNO 3 4 g. C in (HCO 3 ...

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Modern Chemistry Chapter 7 Review Answers Section 3

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7 Chemical Formulas and Chemical Compounds

CHAPTER 5 REVIEW The Periodic Law SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. c In the modern periodic table, elements are ordered (a) according to decreasing atomic mass. (b) according to Mendeleev's original design. (c) according to increasing atomic number. (d) based on when they were discovered. 2. d Mendeleev noticed that certain similarities in the ...

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CHAPTER Section 6.3 continued Date Class STUDY GUIDE FOR CONTENT MASTE in your textbook, read about ionization energy and electronegativity. Answer the following questions. 7. What is ionization energy? 8. Explain why an atom with a high ionization-energy value is not likely to form a positive ion. 9.

Prentice Hall World History: The Modern Era, Author ...

CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. Write formulas for the following compounds: CuCO 3 a. copper(II) carbonate Na 2SO 3 b. sodium sulfite (NH 4) 3PO 4 c. ammonium phosphate SnS 2 d. tin(IV) sulfide HNO 2 e. nitrous acid 2.