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Access Answers to NCERT Exemplar Class 11 Chemistry Chapter 1. Multiple Choice Questions (Type-1) 1. Two students performed the same experiment separately and each one of. them recorded two readings of mass which are given below. The correct reading. of mass is 3.0 g. Based on given data, mark the correct option out of the. following statements.

11.1 The Dissolution Process – Chemistry
Now let's move on to the most common types of nuclear decay. The first is alpha decay.In alpha decay, the nucleus emits an alpha particle, or a particle containing two protons and two neutrons.

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Download eSaral App for Video Lectures, Complete Revision, Study Material and much more... Sol. (i) Average atomic mass : It is defined as average of the mass of all of the atoms of an elements, e.g., average atomic mass of is 35.5 u. (ii) Mole is defined as amount of substance that contains as many atoms, molecules and particles as there are atoms in exactly 0.012 kg of Carbon-12 isotope.

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The quantum mechanical model of the atom, or planetary model, visualizes nuclei and electron orbitals similarly to planets orbiting a sun. Learn more about the definition of the quantum mechanical ...

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An earlier chapter of this text introduced solutions, defined as homogeneous mixtures of two or more substances.Often, one component of a solution is present at a significantly greater concentration, in which case it is called the solvent.The other components of the solution present in relatively lesser concentrations are called solutes.Sugar is a covalent solid composed of sucrose molecules ...

Class 11 Chemistry Revision Notes for Chapter 7 - Equilibrium
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Class 11 Chemistry chapter equilibrium notes explain different features of equilibrium state as – (1) density, pressure, concentration, colour remaining constant, (2) achieved only within a closed vessel, (3) nature of the state is reversible, (4) nature of the state is dynamic, i.e., the system remains in motion, and (4) rate of forward and the backward reaction is equal.